#### **ELECTROCHEMICAL SYNTHESIS**

### OF NEW 2-(2'-FURYL)BENZ-

#### **IMIDAZOLE DERIVATIVES**

## A. A. Aleksandrov, T. G. Galkin, M. M. El'chaninov, and O. V. Popova

**Keywords:** 2-(2',5'-dimethoxy-2',5'-dihydrofuryl)benzimidazole, 2-(2',5'-dimethoxy-2',5'-dihydrofuryl)-1-methylbenzimidazole, 2-(2'-furyl)benzimidazole, *cis* and *trans* isomers, electrochemical methoxylation.

Neither the chemical nor electrochemical methoxylation of furan derivatives of benzimidazole has been studied extensively. We have found that the introduction of a substituent such as 2-benzimidazolyl at  $C_{(2)}$  of the furan ring does not hinder the methoxylation reaction as previously observed for bis(aminoalkyl)furans [1].

2-(2'-Furyl)benzimidazole (1) and its N-methylation product, 2-(2'-furyl)-1-methylbenzimidazole (2) [2] were subjected to electrolysis. The 2',5'-dimethoxy-2',5'-dihydrofuryl derivatives obtained in both cases are ~70:30 mixtures of two geometrical isomers as indicated by <sup>1</sup>H NMR spectroscopy [3].

1a, 1b Bzm = 2-benzimidazolyl 2a, 2b Bzm = 1-methyl-2-benzimidazolyl

The isomers obtained from **2** were isolated as pure compounds and their structure was established by <sup>1</sup>H NMR spectroscopy in CDCl<sub>3</sub>. Thus, **2a** is likely the *cis* isomer since the singlet for the furan 5'-H proton is downfield relative to the corresponding signal in the spectrum of **2b**. This shift is undoubtedly related to the effect of the benzimidazolyl group. The methoxylation product of **1** could not be separated into isomers.

2-(2',5'-Dimethoxy-2',5'-dihydro-2'-furyl)benzimidazoles 1a and 1b, Isomer Mixture. A sample of 1 (3.68 g, 20 mmol) and ammonium bromide (1.96 g, 20 mmol) were dissolved in methanol (50 ml). The solution was placed in an electrolyzer with a platinum anode and nickel cathode and cooled to -15°C. The temperature was maintained during the experiment from -5 to -15°C. The electrolysis was carried out with anodic current density  $0.05 \text{ A/cm}^2$  until  $2.90 \cdot 10^5$ - $3.86 \cdot 10^5$  C/mol (3-4 F/mol) was passed. The reaction mixture was neutralized by adding aqueous ammonia. Methanol was distilled off in vacuum using a water pump. The residue was extracted with  $CH_2Cl_2$ , dried over  $Na_2SO_4$ , and subjected to chromatography on a 100-cm column

South Russian State Technical University, 346428 Novocherkassk, Russia; e-mail: andrey@srstu.novoch.ru. Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 8, pp. 1131-1132, August, 2001. Original article submitted April 23, 2001.

(d = 2.5 cm) packed with 300 g of alumina (Brockmann grade II activity) using CH<sub>2</sub>Cl<sub>2</sub> as the eluent. Yield of the isomer mixture 4.23 g (86%); mp of the picrate 232-233°C. Found, %: C 48.41; H 3.87; N 15.08. C<sub>19</sub>H<sub>17</sub>N<sub>5</sub>O<sub>10</sub>. Calculated, %: C 48.01; H 3.60; N 14.73.

2-(2',5'-Dimethoxy-2',5'-dihydro-2'-furyl)-1-methylbenzimidazoles (2a) and (2b) were obtained as an isomer mixture by analogy to 1a and 1b from 2 (3.96 g, 20 mmol). After chromatography, the following indices were obtained for the individual isomers.

*cis* Isomer 2a was obtained in 46% yield (1.82 g); mp 64-65°C (pentane). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>), δ, ppm, J (Hz): 3.10 (3H, s, 2'-OMe); 3.28 (3H, s, 5'-OMe); 3.96 (3H, s, N–CH<sub>3</sub>); 6.10 (1H, s, 5'-H); 6.25 (1H, d, J = 5.9, 4'-H); 7.15-7.50 (3H, m, Ar); 7.73 (1H, d, J = 8.3, Ar). Found, %: C 65.03; H 6.09; N 11.06. C<sub>14</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>. Calculated, %: C 64.60; H 6.20; N 10.76.

*trans* Isomer 2b was obtained in 15% yield (0.59 g); mp 68-69°C (heptane). <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>), δ, ppm, J (Hz): 3.37 (3H, s, 2'-OMe); 3.53 (3H, s, 5'-OMe); 3.96 (3H, s, N–CH<sub>3</sub>); 5.58 (1H, s, 5'-H); 6.25 (1H, d, J = 5.9, 3'-H); 6.58 (1H, d, J = 5.9, 4'-H); 7.15-7.50 (3H, m, Ar); 7.73 (1H, d, J = 8.3, Ar). Found, %: C 64.8; H 5.82. C<sub>14</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub>. Calculated, %: C 64.60; H 6.20.

# **REFERENCES**

- 1. K. Yu. Novitskii, N. K. Sadovaya, and Yu. K. Yur'ev, Khim. Geterotsikl. Soedin., 826 (1966).
- 2. M. M. El'chaninov, L. Ya. Oleinikova, and A. M. Simonov, *Khim. Geterotsikl. Soedin.*, 1047 (1979).
- 3. L. N. Kralinina, R. I. Kruglikova, V. V. Yastrebov, G. P. Krutetskaya, A. I. Chernyshov, and G. I. Samokhvalov, in: *Heteroorganic Compounds, Vol. 2, Oxygen-Containing Heterocycles* [in Russian], Riga (1970), p. 6.